Name
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# From Notes To Report Answer Key

# Sample Rewrite

During the experiment, the reactants-baking soda (sodium bicarbonate) and vinegar (acetic acid)-underwent a chemical reaction that produced carbon dioxide gas, causing the balloon to inflate. The container became cool to the touch, indicating an endothermic reaction. Observations confirmed that the gas released was carbon dioxide, similar to that produced by human respiration. The reaction occurred rapidly, demonstrating how combining an acid and a base results in gas production and energy absorption.

#### Teacher's Guide

 Skill Focus: Students practice transforming informal observations into formal technical writing, reinforcing their command of discipline-specific vocabulary and scientific structure.

## • Differentiation Tips:

- o For emerging writers, provide a model outline (introduction  $\rightarrow$  method  $\rightarrow$  observation  $\rightarrow$  conclusion).
- For advanced students, encourage inclusion of chemical equations or additional terminology such as reaction rate or energy transfer.

# Engagement Ideas:

- Have students perform a brief, safe classroom demonstration (e.g., vinegar and baking soda) before rewriting.
- Use peer review: students exchange rewritten reports and highlight each other's effective use of technical terms.

### Extension Activities:

- Ask students to rewrite a field note from another subject (e.g., geography or environmental science) into a professional report.
- Challenge advanced learners to turn their field report into a short presentation or infographic using proper technical vocabulary.

