Name			

Float or Sink? The Science of Buoyancy

Whether an object floats or sinks depends on its density compared to water.

$$D = \frac{m}{V}$$

If $D_{\text{object}} < D_{\text{water}} \rightarrow \text{it floats}$ or If $D_{\text{object}} > D_{\text{water}} \rightarrow \text{it sinks}$

Water's density: $1.0 \text{ g/mL} = 1000 \text{ kg/m}^3$

The **fraction of an object submerged** in water equals:



- 1. A cube of pine wood has a mass of 120 g and a volume of 200 cm³. Will it float or sink in water?
- 2. An aluminum block has a mass of 540 g and a volume of 200 cm³. Does it float or sink?
- 3. An ice cube has a mass of 9.2 g and a volume of 10 cm³. What is its density? Will it float in water?
- 4. A boat has a total volume of 400 m³ and a mass of 280,000 kg. Find its average density in kg/m³ and determine if it will float.
- 5. A rock has a density of 2.8 g/cm³. How many times denser is it than water?
- 6. An object has a density of 0.8 g/cm³. What fraction of its volume will be underwater when it floats?
- 7. A student drops three objects into a tank of water:

Object A: 0.6 g/mL Object B: 1.2 g/mL Object C: 1.0 g/mL

Which float, sink, or remain suspended?

- 8. The density of seawater is 1.03 g/mL. An object has a density of 1.00 g/mL. Will it float higher, lower, or the same as it would in freshwater? _____
- 9. A piece of wax has a mass of **24 g** and displaces **30 mL** of water. What is its density, and does it float?
- 10. Challenge Question: A ship floats with **80% of its volume underwater**. What is the ship's average density in g/mL?

